



## Product Data Sheet

### AmberSep™ GT74 Chelating Resin

Industrial-grade Complexing Resin

#### Description

AmberSep™ GT74 Chelating Resin is a weakly acidic cation exchange resin with very pronounced selectivity for certain metal ions such as mercury, rhodium, copper, silver, cadmium, and lead.

AmberSep™ GT74 has been developed for the removal of mercury from different solutions and gaseous streams and can be regenerated very efficiently with hydrochloric acid.

AmberSep™ GT74 is insoluble in common solvents and stable over the entire pH range. Oxidizing media should be avoided. The special properties of AmberSep™ GT74 can be useful for problems where removal of metal ions Cu, Ag, Pb, and Cd is desired. Applications may be found in different fields of chemical technology such as wastewater treatment, recovery of solutions and metals in the plating industry, recovery of catalysts and removal of interfering ions in hydrometallurgy.

#### Applications

- Wastewater treatment
- Flue gas desulfurization blowdown
- Electroplating
- Hydrometallurgy
- Chlor-alkali streams for mercury cell electrodes

#### Typical Properties

<b>Physical Properties</b>	
Copolymer	Styrene-divinylbenzene
Matrix	Macroporous
Type	Chelant
Functional Group	Thiol
Physical Form	Green-gray, opaque, spherical beads
<b>Chemical Properties</b>	
Ionic Form as Shipped	H <sup>+</sup>
Total Exchange Capacity	≥ 1.40 eq/L (SH form)
Water Retention Capacity	38.0 – 46.0% (H <sup>+</sup> form)
<b>Particle Size</b> §	
Particle Diameter	450 – 700 μm
Uniformity Coefficient	≤ 1.60
< 425 μm	≤ 12.0%
> 850 μm	≤ 10.0%
<b>Density</b>	
Bulk Density as Shipped	784 g/L

§ For additional particle size information, please refer to the [Particle Size Distribution Cross Reference Chart](#) (Form No. 45-D00954-en).

## Suggested Operating Conditions

Maximum Operating Temperature	60°C (140°F)
pH Range	0 – 14
Bed Depth, min.	1000 mm (3.1 ft)
Flowrates	
Service	10 BV*/h (1.25 gpm/ft <sup>3</sup> )
Backwash	About 12 m/h (5 gpm/ft <sup>2</sup> ) with water at 20°C (68°F)
Total Rinse Requirement	2 – 3 BV (15 – 22.5 gal/ft <sup>3</sup> )
Regenerant	Concentrated HCl

\* 1 BV (Bed Volume) = 1 m<sup>3</sup> solution per m<sup>3</sup> resin or 7.5 gal per ft<sup>3</sup> resin

## Application Information

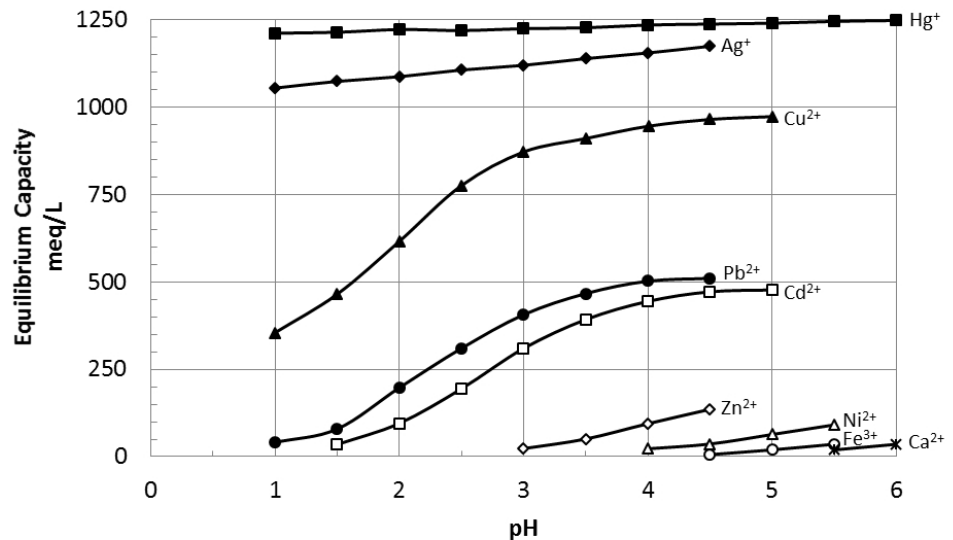
### Selectivity Sequence

The selectivity sequence of AmberSep™ GT74 Chelating Resin is:

Hg > Ag > Cu > Pb > Cd > Ni > Co > Fe > Ca > Na

### Equilibrium Capacity Data

The high selectivity of AmberSep™ GT74 Chelating Resin for certain metals is shown in the graph below as a function of pH. All data were determined in a normal solution of NaNO<sub>3</sub>. The resin has a pronounced preference for copper, lead, and cadmium ions, which are removed in considerable quantities, even from solutions containing only 1 meq/L of metal and a large excess of Na<sup>+</sup> ions. The data indicate the possibility of selective separation of these metals.



## Application Information (Cont.)

### Example: Removal of lead from wastewater

Influent composition:

Pb <sup>2+</sup>	6 ppm
Sb <sup>3+</sup>	0.3 ppm
Na <sup>+</sup>	100 ppm
pH	2.5

In the experiment, the solution passed through a column of AmberSep™ GT74 Chelating Resin at a flowrate of 15 m/h (6 gpm/ft<sup>2</sup>). The effluent contained less than 0.01 ppm Pb. After passage of 700 bed volumes of the solution, the effluent composition was still unchanged.

## Product Stewardship

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Please be aware of the following:

- **WARNING:** Oxidizing agents such as nitric acid attack organic ion exchange resins under certain conditions. This could lead to anything from slight resin degradation to a violent exothermic reaction (explosion). Before using strong oxidizing agents, consult sources knowledgeable in handling such materials.

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